Reading Questions		Name	
Solutions and Other Mixtures Pp. 186-190			
1. What are the th	aree objective of this section?		
2. How are mixtu	res different from substances?		
2 377	0		
3. What is a heter	rogeneous mixture?		
4. What is a suspe	ension?		
5. How big are pa	articles in a suspension?		
0 1	•		
6. What is a collo	:49		
o. What is a cono	olu !		
7. Why doesn't a	colloid settle?		
8. What are immi	scible liquids?		
9. How do you m	ake mayonnaise?		
2. 110 do you iii			
10. What is an emulsion?			

11. What is a homogenous mixture?

12. What does uniform mean?

13. What is a solution?

14. What are the parts of a solution?
15. Give examples of solutions that are not solids in a liquid?
16. What is miscible?
17. Why is it hard to separate miscible liquids?
18. How does distillation work?
19. What is chromatography?